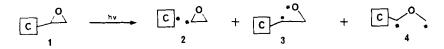
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PHOTOCHEMICAL ACTIVATION OF  $\alpha$ - $\beta$  EPOXYKETONES. FORMATION OF BENZANNELATED PERHYDROAZULENES BY INTRAMOLECULAR TRAPPING OF A REACTIVE INTERMEDIATE

Ken S. Feldman<sup>†</sup> Chemistry Department, Stanford University, Stanford, California 94305

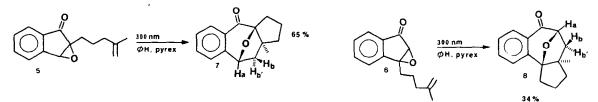
Summary: Linear and angular benzannelated perhydroazulenes can be efficiently prepared by intramolecular cycloaddition of unactivated oletins to reactive intermediates generated by irradiation of  $\alpha - \beta$  epoxyketones.

Excitation of chromophores containing adjacent oxiranes  $\frac{1}{\sqrt{2}}$  are known to produce several types of high energy reactive intermediates corresponding to the different modes of  $\sigma$ -bond cleavage.<sup>1</sup>



The reactive intermediate 4 has been shown to undergo addition reactions with suitably activated olefins or carbonyl compounds to generate tetrahydrofurans or 1,3 dioxolanes.<sup>1</sup> However, the electronic nature of 4 (triplet diradical \*\* = ... or carbonyl ylid \*\* = +, -) remains an unsettled issue.

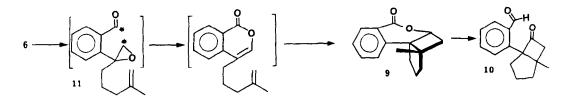
Herein we report that irradiation of aromatic epoxy ketones 35 and 6 produce intermediates (such as 4) than can be efficiently trapped in an intramolecular manner by unactivated olefins. The assignment of gross structure 7 and 8 depend upon comparison of their spectral data with that of carbonyl ylid



adducts of known structure.<sup>4</sup> The stereochemical assignment of 7 is based on the observed coupling constants (7,  $J_{ab} = 7.0 \text{ Hz}$ ,  $J_{ab}$ , = 0 Hz). The stereochemical assignment of 8 is again based on the observed coupling constants (8,  $J_{ab} = 8.7 \text{ Hz}$ ,  $J_{ab}$ , = 1.3 Hz) and the position of the angular methyl signal (8,  $\delta$  0.65, compare 7,  $\delta$  0.84). Examination of molecular models suggests that the methyl group in 8 is positioned almost directly over the aromatic ring and should be shielded relative to that in 7. In addition, when the methyl signal of 8 is irradiated an 11% nuclear Overhauser enhancement of H<sub>b</sub>, (and also small enhancements of the aromatic protons) is observed.

In addition to g, irradiation of  $f_0$  also produces another compound in equal amounts (35%) which is assigned the structure g based on spectral data: (IR:

1717  $cm^{-1}$ ; irradiation of the angular methyl induces a nuclear Overhauser enhancement of a cyclobutane proton and an aromatic proton ( $\delta$ 7.15, d, J = 7.88 Hz)).



Further, reduction of 2 (LAH) leads to a diol (ms:m/e 232, NMR (CDCl<sub>3</sub>,D<sub>2</sub>O)  $\delta$ 4.59 (d, J = 11.5 Hz, ArC(H)HOD), 4.39 (d, J = 11.5 Hz, ArCH(H)OD), 4.29 (dd, J = 6.4, J)1.2 Hz, C(H)OD)) which can be oxidized under Swern's conditions to yield the ketoaldehyde  $10_{0.0}$  (IR: 1770 cm<sup>-1</sup>, cyclobutanone; NMR (CDCl<sub>3</sub>)  $\delta$  10.18 (s, ArCHO)). A similar reduction/oxidation cycle with 8 simply regenerates 8.

The obtention of  $\delta_{\mathcal{A}}$  and  $\theta_{\mathcal{A}}$  suggests a mechanistic rationale in which irradiation of 6 leads to roughly equal partitioning through intermediates of types 2 (cf. 11) and 4. 11 then undergoes rearrangement to an isocoumarin<sup>1f</sup>, the direct precursor of 9 in a second photochemical process. Irrespective of the mechanistic course of these reactions, the tetracycles 7 and 8 embody the gross skeletal structures of diterpenes of the dolastane<sup>5</sup> and daphnane<sup>6</sup> classes. These transformations may serve as useful routes to these complex natural products. Acknowlegement: I thank Dr. E. E. van Tamelen for his financial support, NSF Grant Che-81-09064 for NMR facilities, and Charles Manly for his assistance in obtaining and interpreting the NOE data.

† Address correspondence to: Chemistry Department, 152 Davey Laboratory, Pennsylvania State University, University Park, PA 16802

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